

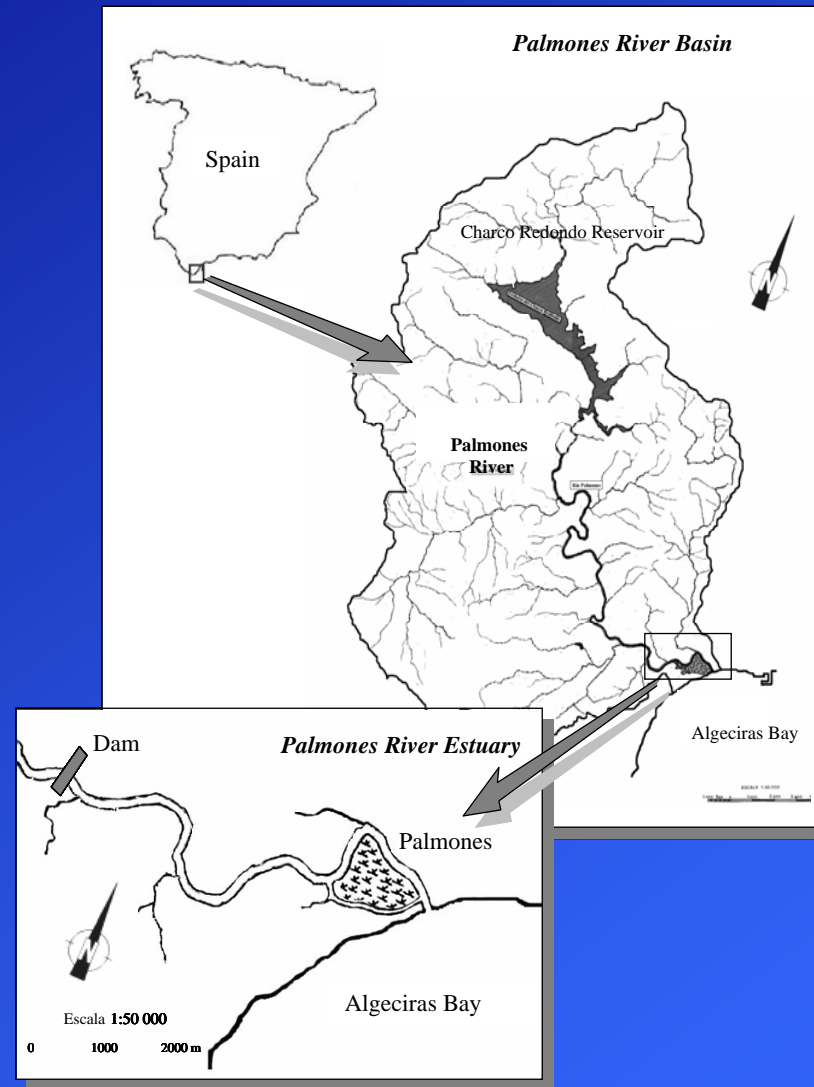


UNIVERSITY OF MÁLAGA
Faculty of Sciences
Department of Ecology and Geology
Spain

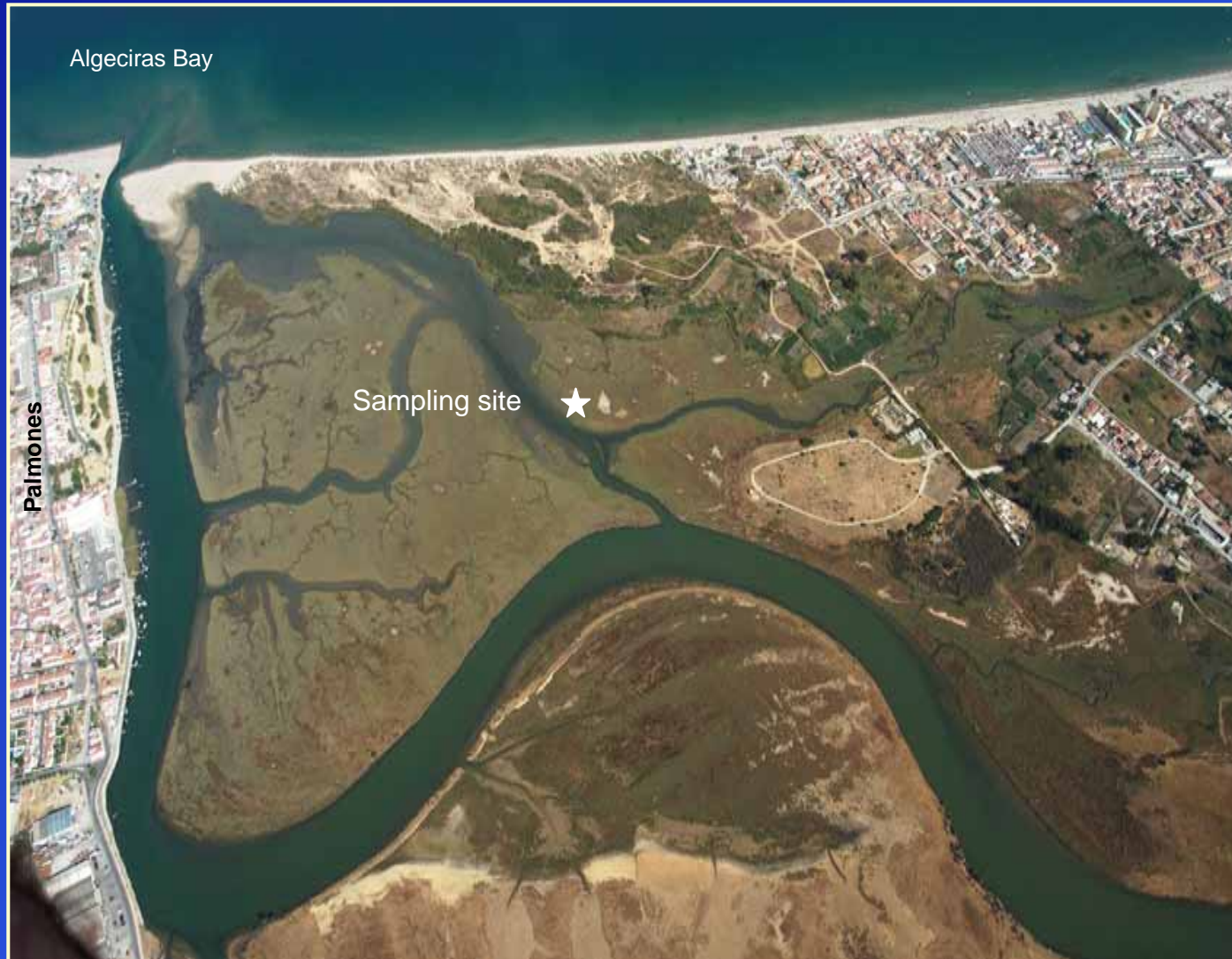
Temperature Methanogenesis Regulation in Shallow Temperate Estuaries

Sonia Moreno and F. Xavier Niell





Palmones River Estuary (Spain)



Palmones River Estuary (Spain)

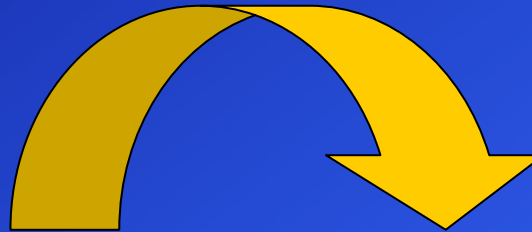


Methanogens (*Archaea*)



**Organic matter
mineralization**

$\Delta G^{\circ} = - 31 \text{ KJ}$



Organic C

(acetate, methanol, ethanol...)

Inorganic C + CH₄

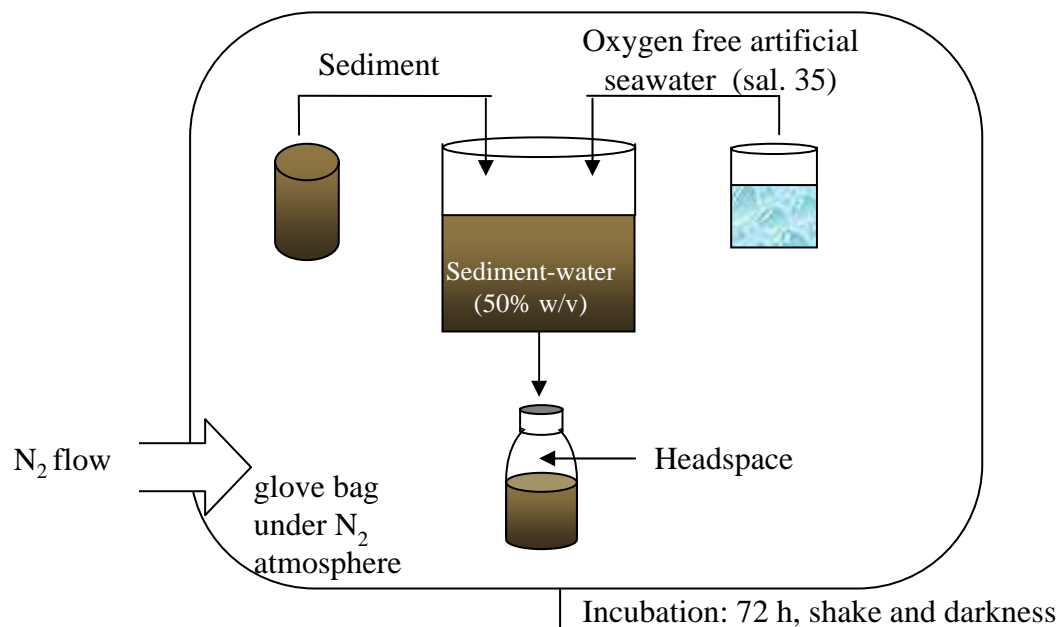
Anaerobic environment

Methane was measured into the sediment and onto it

- **PRODUCTION** is the difference between the methane concentration measured between two times. **Slurry**
- **EMISION** is the amount of methane degaged from sediment during a period of time. **Core**

- **Methane production rate** was determined using the **slurry technique** described by Smith et al (1978).

Slurry technique described by Smith et al (1978)

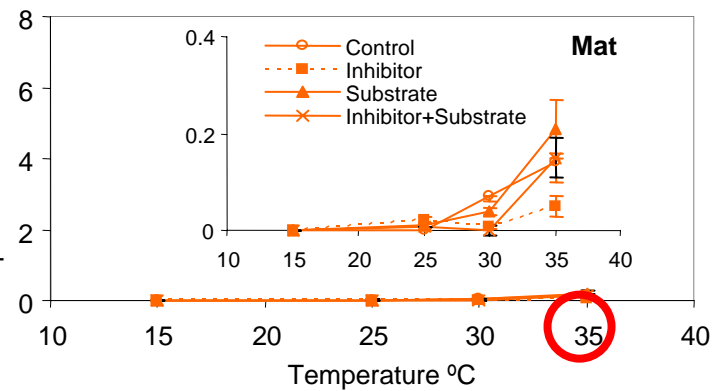
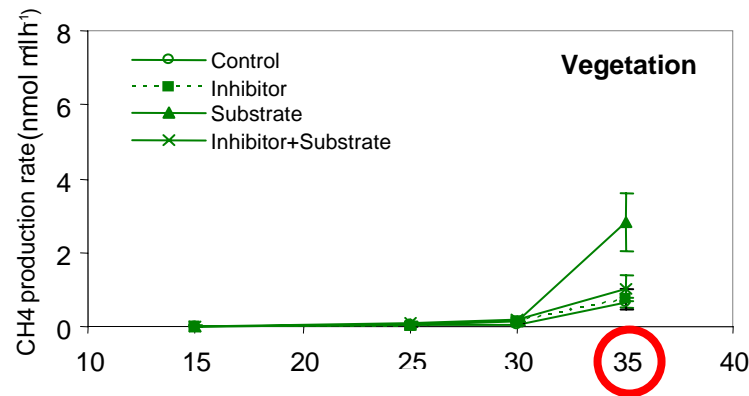
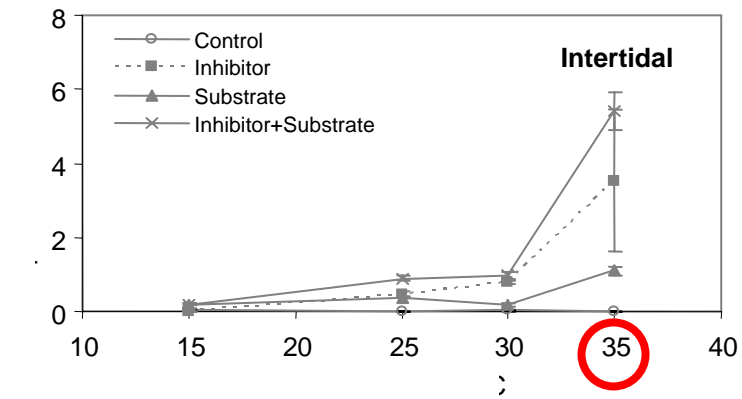


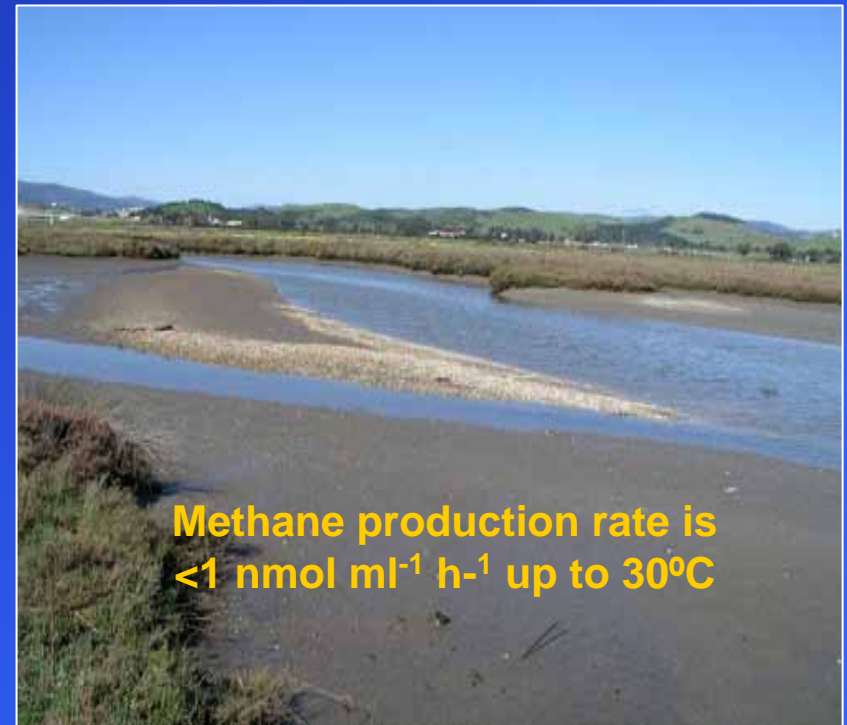
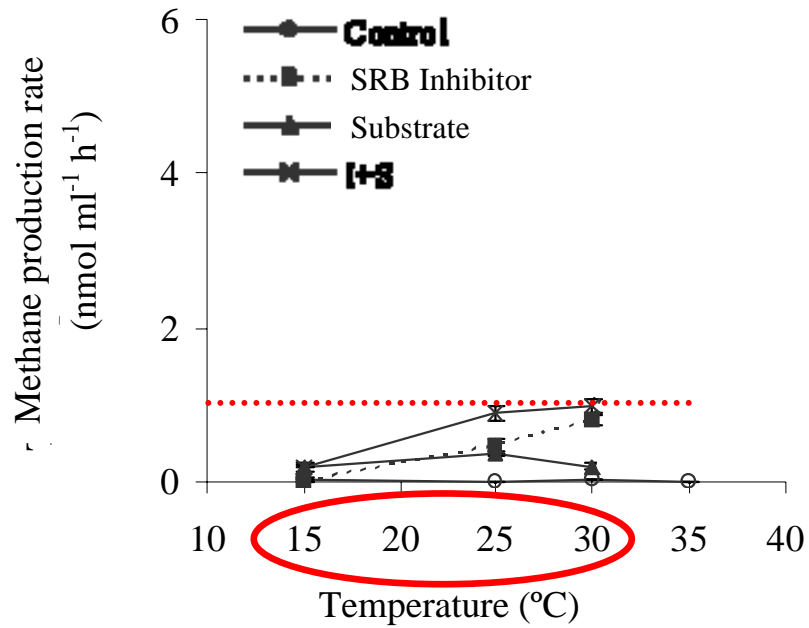
Temperature	Substrate	Competition with SRB	Sulphahte inhibition	Salinity inhibition
15, 25, 30 y 35° C Treatments: •Control •Substrate: acetate •Inhibitor: molybdate •Substrate+Inhibitor Sediment: •Intertidal •Vegetation •Mat	35° C Treatments: •Control •Substrate 1: acetate •Substrate 2: methanol Sediment: •River •Intertidal •Vegetation •Mat	35° C Treatments: •Control •Substrate 1: acetate •Substrate 2: methanol •Inhibitor: molybdate •Substrate 1+Inhibitor •Substrate 2+Inhibitor Sediment: River, Intertidal, Vegetation, Mat	35° C Treatments: •Control •SO ₄ Na ₂ (0.1, 1, 5, 25, 50 y 100 mM) Sediment: •River	35° C Treatments: •Control •Gradient of salinity (0, 10, 20, 25, 30 y 35) Sediment: •Palmones River Estuary (Spain) •Tagus River Estuary (Portugal)

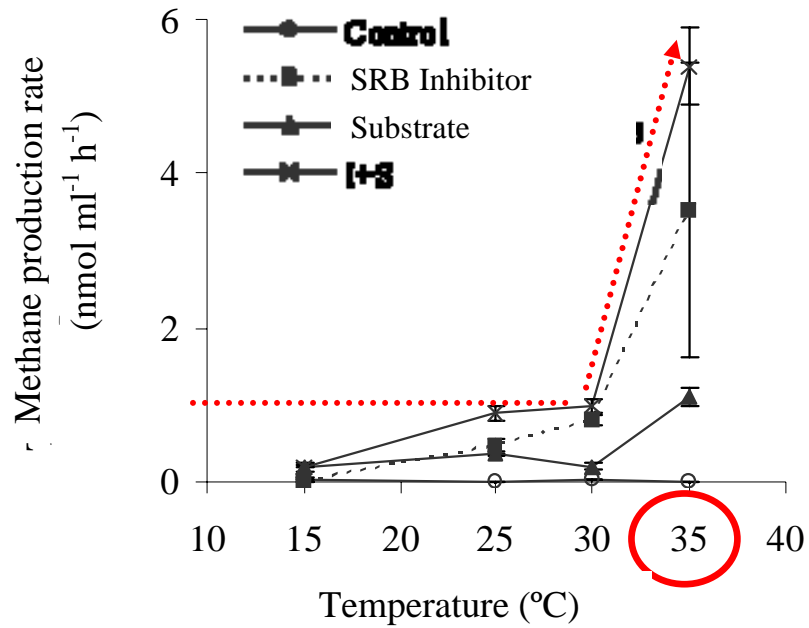
- **Methane production rate** was determined using the **slurry technique** described by Smith et al (1978).
- **Methane concentration** was determined by **gas chromatography**.

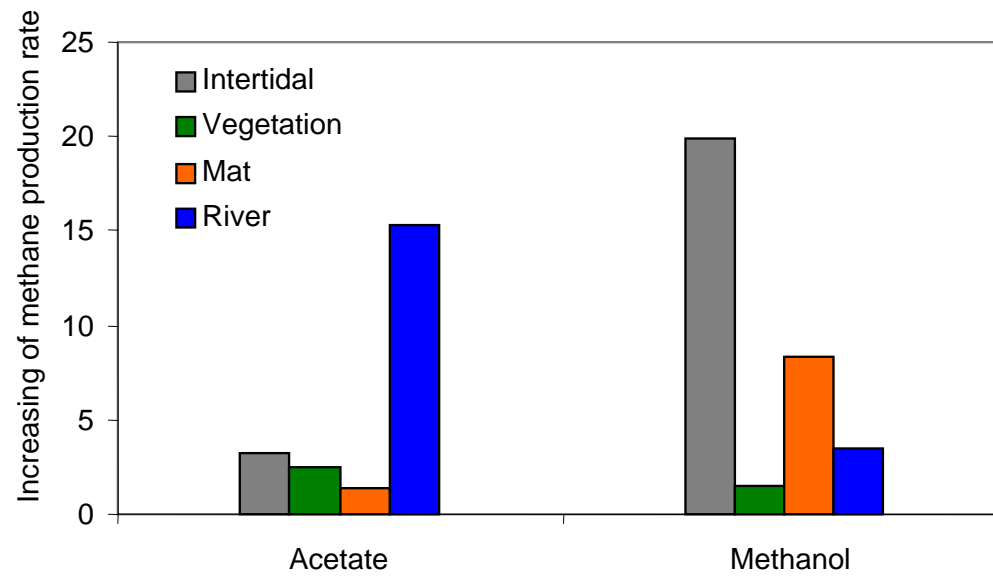


Temperature Effect

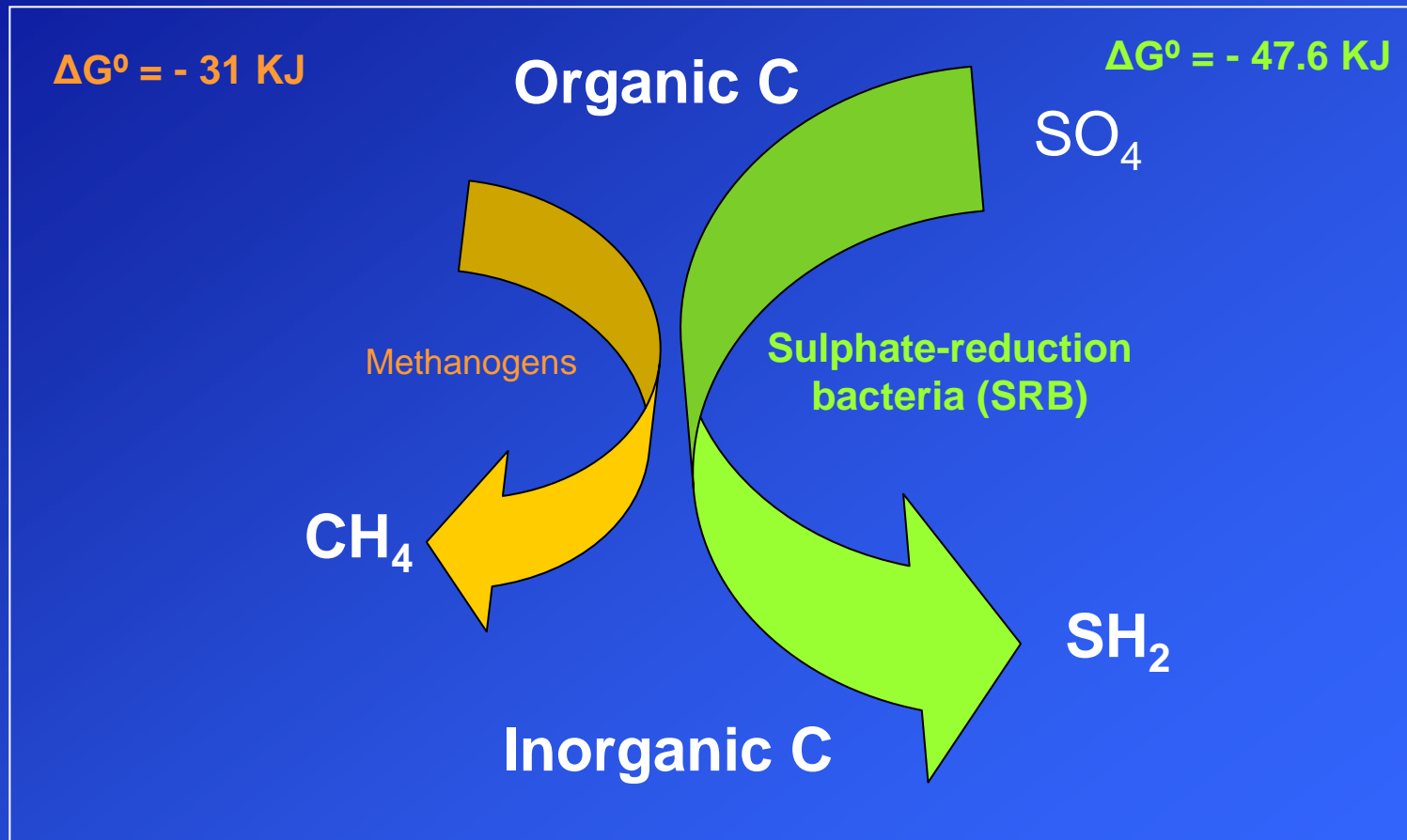






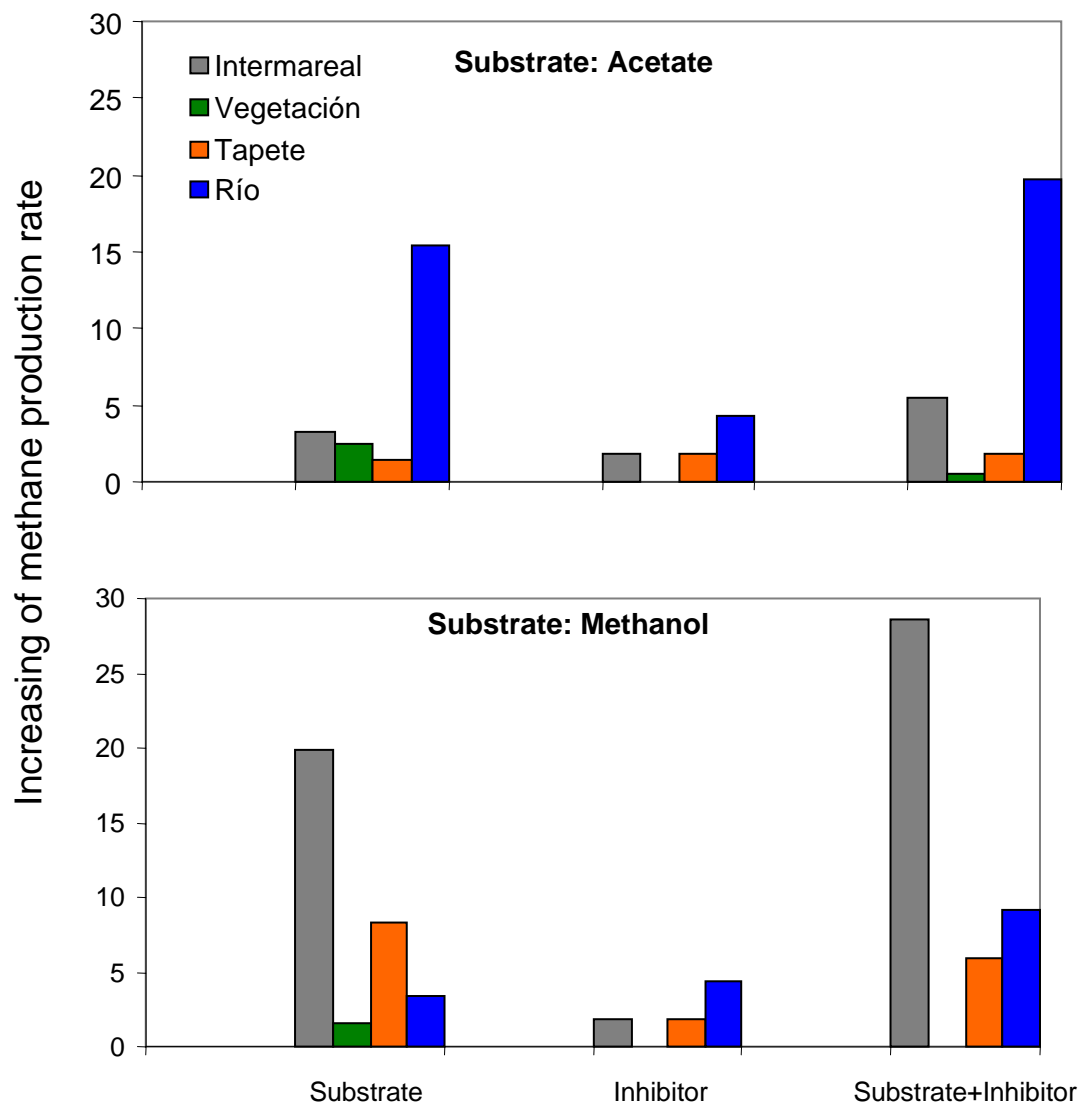


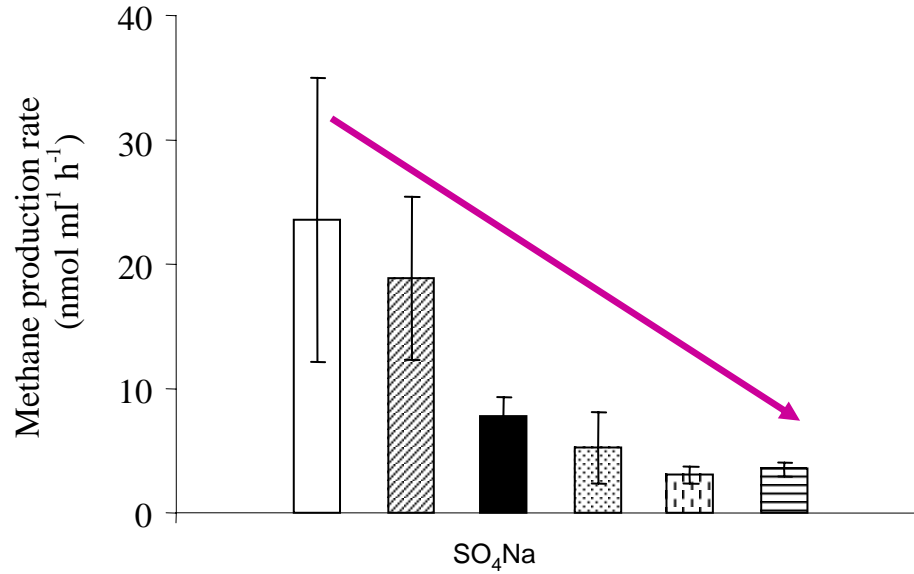
ORGANIC MATTER MINERALIZATION



Rich-sulphate environment

Anaerobic environment

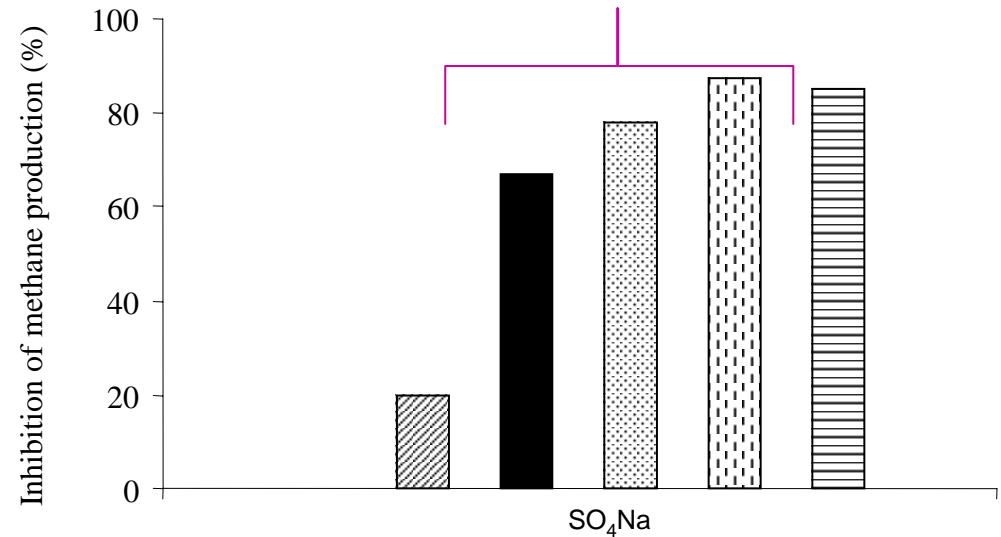




Inhibition of methane production
50 - 90%

SO₄Na concentration in Palmones (5-50 mM)

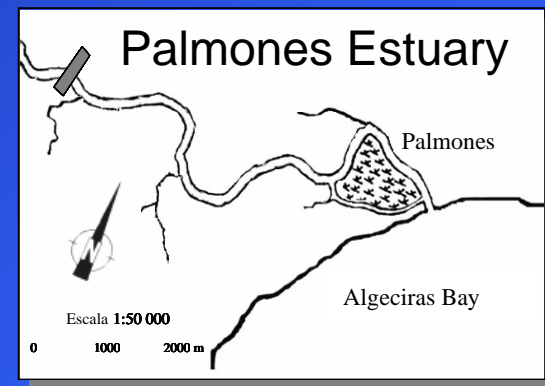
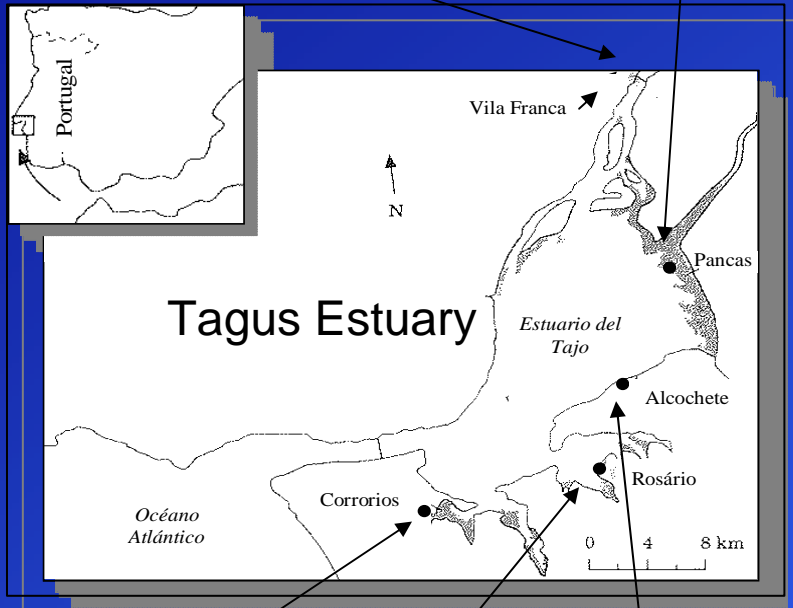
- 0 mM SO₄Na
- ▨ 0.1 mM
- 5 mM
- ▤ 25 mM
- ▥ 50 mM
- ▧ 100 mM



Sal. 0



Sal. 25



Sal. 35



Sal. 30

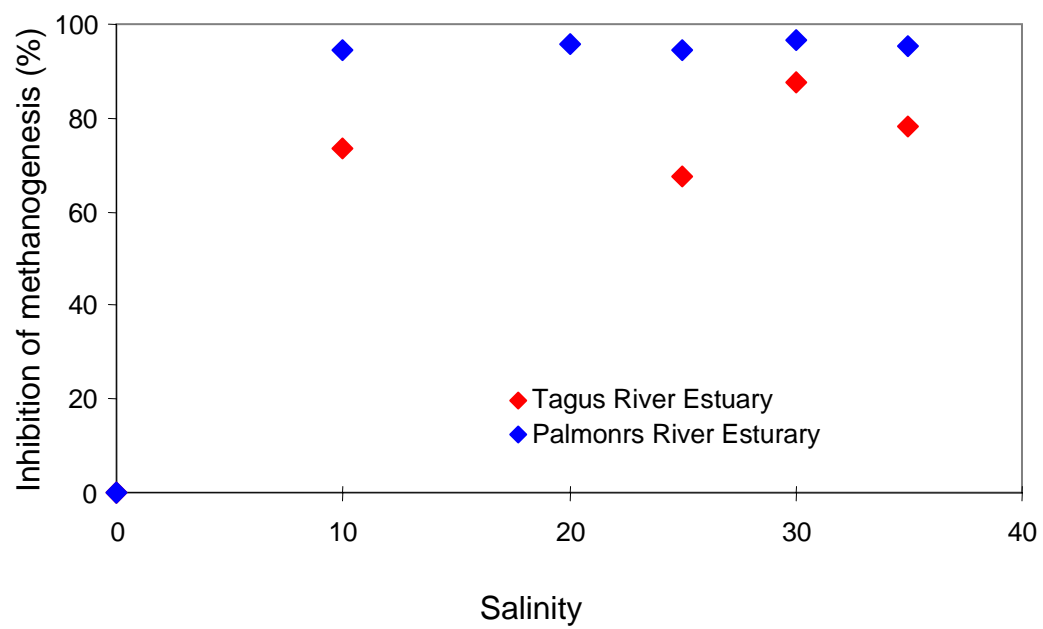


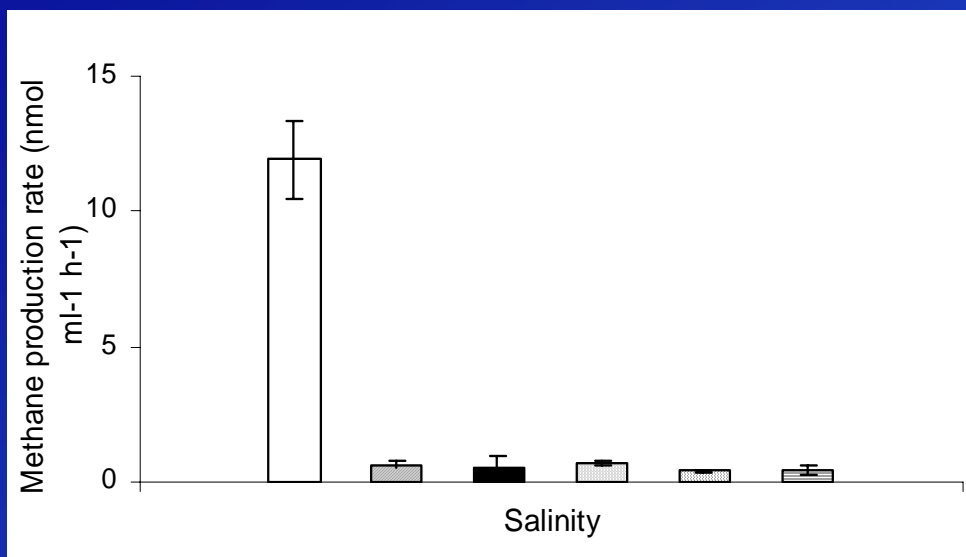
Sal. 10



Sal. 35



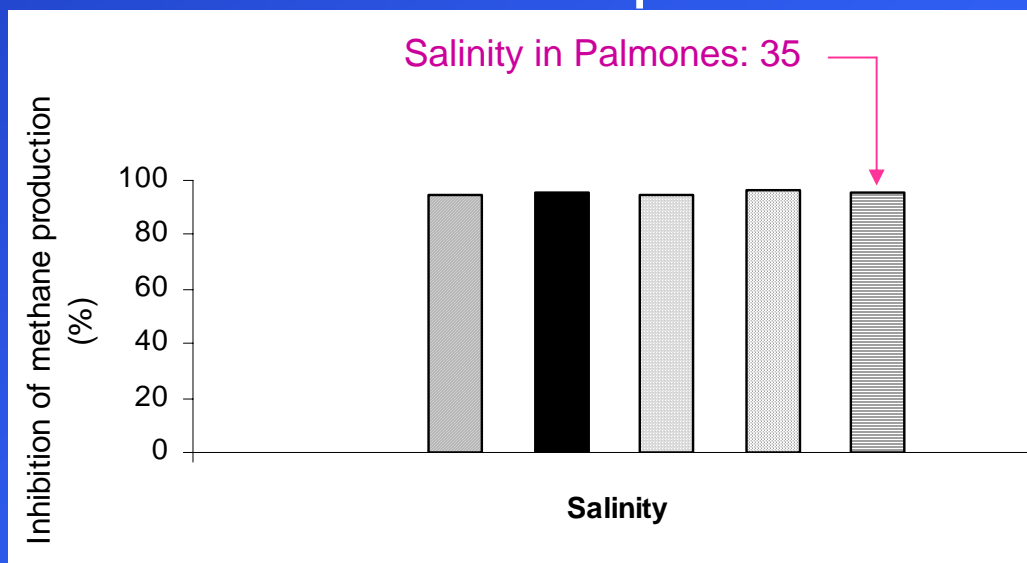




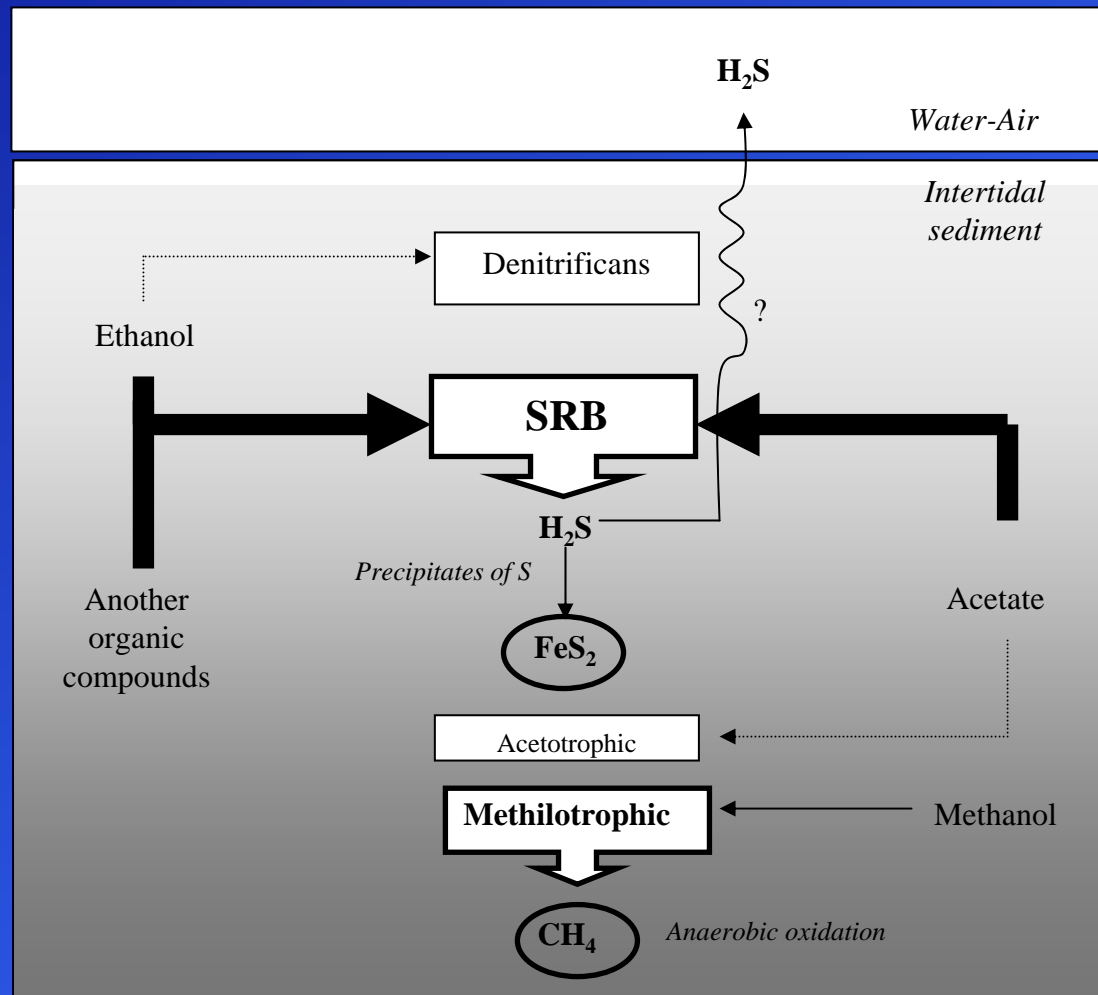
Inhibition of methane production
95%



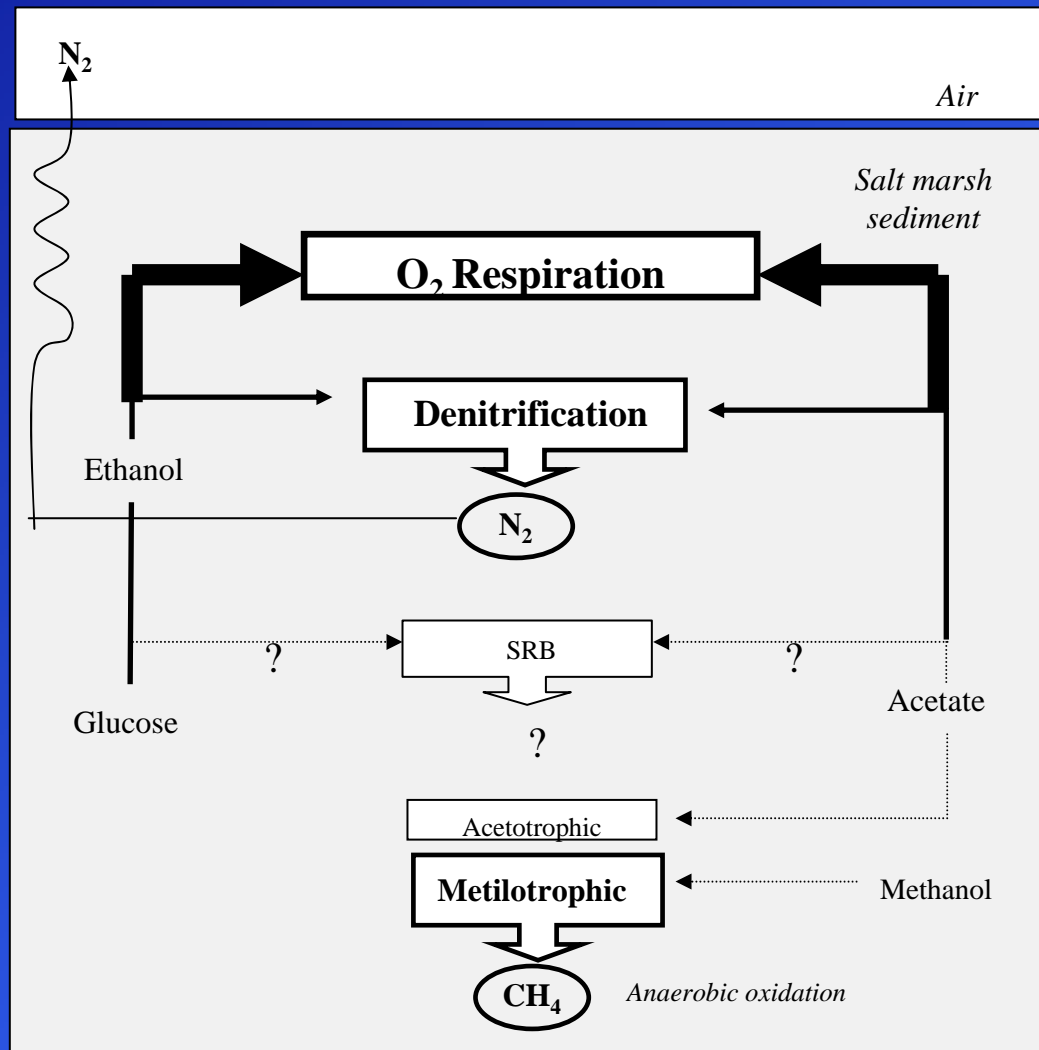
- Control
- 10
- 20
- 25
- 30
- 35



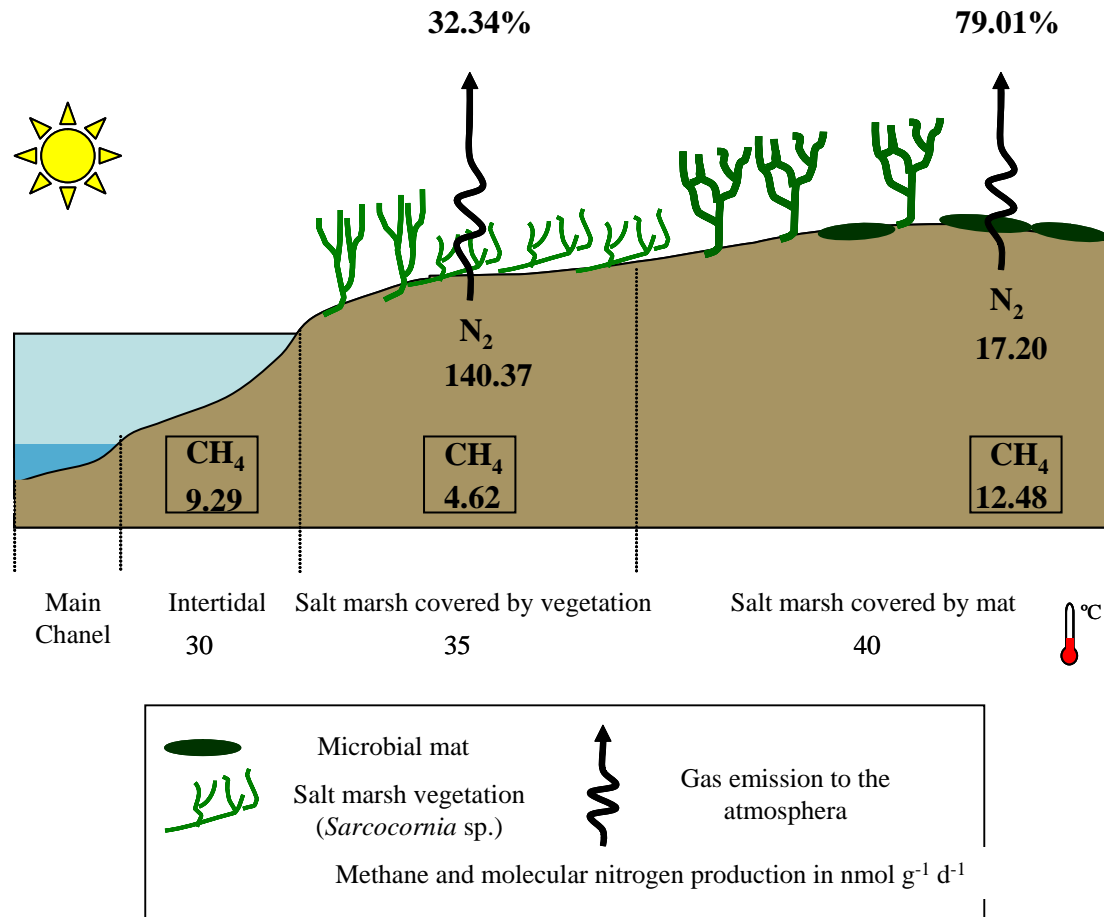
The Mineralization Processes Implicated in Gas Dynamics in the Intertidal Sediment



The Mineralization Processes Implicated in Gas Dynamics in the Salt Marsh Sediment



Gas Production and Emission *in situ* (Summer)

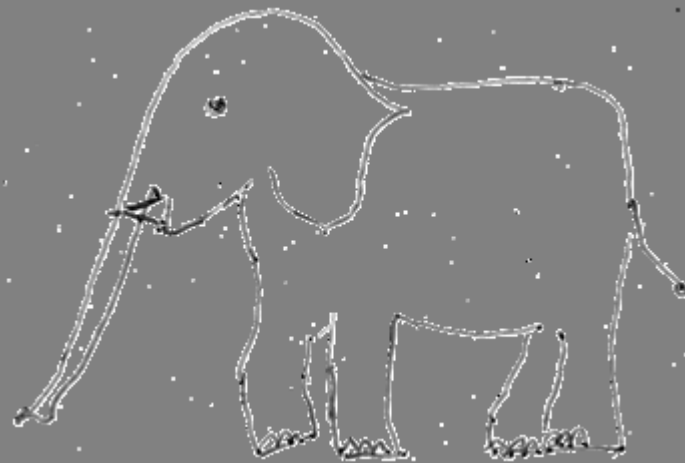


Site	Ecosystem	Observation	Production (nmol ml ⁻¹ h ⁻¹)
Watson and Nedwell (1998) Ellergrower Moss (New Galloway, Escocia)	Peat	“hollow” peat <i>In situ</i> temperature :14.9° C	1-6.8
Great Dun Fell (Cumbria, England)		“wet” peat	0.25-0.43
Furtado and Casper (2000) Dagow lake (Brandenburg, Germany)	Eutrophic lake	<i>In situ</i> temperature: 12° C	1.25
		Temperature: 22° C	6
Present study (2008)	Intertidal	<i>In situ</i> temperature: 30° C	0.58
Palmones River Estuary (Algeciras Bay, Spain)	Salt marsh (vegetation)	<i>In situ</i> temperature: 35° C	0.29
	Salt marsh (mat)	<i>In situ</i> temperature: 40° C	0.79

1. Estuarine sediments in temperate areas (Mediterranean and South European Atlantic) are not a source of methane emission
2. Methane production into the sediment is limited by:
 - Temperature below 30° C
 - Salinity at regular concentrations
 - Organic substrate concentrations
 - Competition with sulphate reduction bacteria
3. Emission has NEVER been detected

There are some differences between text-book explanations
and true behaviour of the systems

IS THIS a HAT?



It isn't

